

## Solubility Enhancement of Atorvast at in Calcium by Solid Dispersion Technique

Pooja Verma\*, R.D Gupta, H.S Lamba, Ayushi Chawla

H.R Institute of Pharmacy, Meerut Road, Morta, Ghaziabad, Uttar Pradesh, India.

Submitted: 20-11-2022

Accepted: 30-11-2022

### ABSTRACT

Hyperlipidemia is the presence of elevated abnormal levels of lipids or lipoproteins in the blood. Hyperlipidemia is regarded as a highly modifiable risk factor for cardiovascular diseases common in elderly patients. The Atorvastatin belongs to BCS class II drug having low solubility and high permeability. In the present study attempt was made to improve solubility and dissolution rate of poorly soluble drug by solid dispersion technique using carrier poloxamer 188. The FDA approved atorvastatin in December 1996. Atorvastatin, a synthetic lipid-lowering agent, is an inhibitor of 3-hydroxy-3-methyl-glutaryl-coenzyme A (HMG-CoA) reductase which catalyzes the conversion of HMG-CoA to mevalonate, an early rate-limiting step in cholesterol biosynthesis. Orally administered Atorvastatin calcium shows low bioavailability due to its low solubility in aqueous and acidic media. The objective of present study was to formulate and optimize solid dispersion tablet containing Atrovastatin calcium to improved solubility. The phase solubility study was adopted for the selection of carriers. Phase solubility revealed that the poloxamer188 was sufficiently able to enhance the aqueous solubility of Atrovastatin calcium. The solid dispersions were prepared by Solvent Evaporation method in different ratios viz. 1:1, 1:1.5, 1:2, 1:2.5, 1:3 & 1:3.5. The prepared solid dispersions were evaluated for physical appearance, solubility study, drug content, and in-vitro dissolution study. Thus the solid dispersion of Atorvastatin calcium tablets, comparable with the innovator formulation developed which could overcome the problem of low solubility. In-vitro dissolution solid dispersion tablet data revealed that ATP4 batch of formulation was optimum due to its better drug release (94.26% at 30 minutes) even after solid dispersion. Tablets of optimized ATP4 batch were kept for stability study at 30 0 C ± 2 0 C/65%RH ± 5% RH and 40 0 C ± 2 0 C/75%RH ± 5% RH. Solid dispersion of Atorvastatin calcium tablets, comparable with the innovator formulation, was successfully developed which could overcome the

problem of low solubility and instability in acidic environment.

**Keywords:** Atorvastatin calcium, solid dispersion, solvent evaporation method, aqueous solubility, dissolution rate.

### I. INTRODUCTION:

Hyperlipidemia is also known as hyperlipoproteinemia because these fatty substances travel in the blood attached to proteins. This is the only way that these fatty substances can remain dissolved while in circulation. Other characteristics of lipoproteins are low density lipoprotein (LDL) and high density lipoprotein (HDL). Excess level of LDL indicates the blockage of arteries, which eventually leads to heart attack. In a Population studies have clearly shown that the greater the risk of heart disease(CVDs) is due to the higher the level of LDL cholesterol. As a result, LDL cholesterol referred to as a bad cholesterol. In contrast, the lower the level of HDL cholesterol, is lead to the greater the risk of coronary heart disease. Hence, HDL cholesterol has been labeled as the good cholesterol.

Cholesterol and other fatty substances combine in the bloodstream and are deposited in the blood vessels to form a material called plaque. The increase in lipids can cause plaques to grow over time, leading to obstructions in blood flow. If an obstruction occurs in the coronary arteries, it could result in a heart attack while an obstruction in the arteries of the brain, could lead to stroke.

The BCS was first contrived in 1995 by Amidon et al. since then it has become a benchmark in the regulation of bioavailability and bioequivalence of oral drug formulation. The characteristic of various BCS classes aids as a guiding tool to improving the efficacy of drug development by proper selection of dosage form and bioequivalence tests, to recommend a class of immediate release (IR) drug products.

Solubility Based Classification Of Drugs As Per United State Pharmacopeia (USP). USP & National formulary lists the solubility of a drug as the number of millilitres of solvents in which 1g of solute will

dissolved. The concept of solid dispersion was originally proposed by Sekiguchi & obi, The term solid dispersion is defined as a group of solid products consisting of at least two different components, generally a hydrophilic matrix & a hydrophobic drugs. The matrix can be either crystalline or amorphous. The drug can be dispersed molecularly, in amorphous particle (clusters) or in crystalline particle.

In the solvent evaporation method, solid dispersion is obtained after the evaporation of solvent from the solution containing a drug and carrier. Some polymers hardly used as carriers in the melting method due to their high melting point can be applied in the solvent method. An important prerequisite of this method is the sufficient solubility of the drug and carrier in a solvent or cosolvent. Finding a suitable non-toxic solvent is sometimes difficult because carriers are hydrophilic whereas drugs are hydrophobic. The main advantage of the solvent method is thermal decomposition of drugs or carriers can be prevented, because of the relatively low temperatures required for the evaporation of organic solvents.

## II. MATERIALS AND METHODS

### Materials:

Atorvastatin calcium was purchased from Psychotropic India Limited, IP-2, Haridwar, Utrakhand. Poloxamer was obtained as a gift sample from Ranbaxy Research and development Center Gurgaon, Haryana & other excipient such as Microcrystalline cellulose, Starch Sodium starch glycolate, Magnesium stearate, & Talc are used.

### Methods:

#### Innovator Tablet Drug Release Profile in 40mg IR tablet

Media and dissolution condition:- The in-vitro release of Innovator IR 40mg tablets was carried out for 30min in pH 6.8 phosphate buffer. The studies were performed in USP dissolution apparatus II at  $37 \pm 0.5^\circ$  C, 75 rpm speed and 900ml volume. Samples were taken at 5min, 10min, 15min, 20min, 25min and 30min and diluted to suitable concentration and analyzed for API content at 246.0 nm by using UV-visible spectrophotometer.

Time(min)	Cumulative%drugrelease
0	0
5	25.32
10	43.71
15	61.20
20	73.80
25	86.90
30	95.65

Table1 Time point innovator's release profile

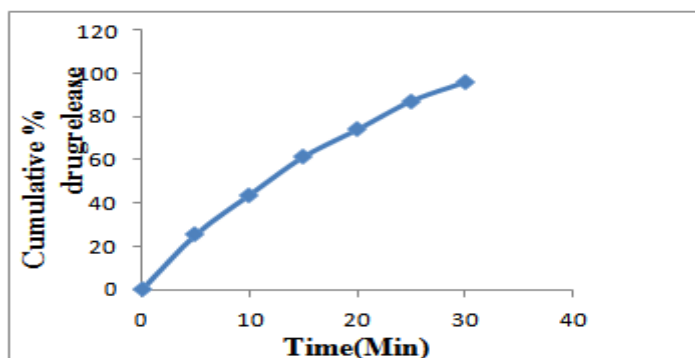


Figure1 Innovator's drug release profile

**Determination of  $\lambda_{max}$**

**(a) Procedure**

Atorvastatin calcium was dissolved in pH 6.8 Phosphate buffer. The solution was scanned for maximum absorbance in UV double beam spectrophotometer [Shimadzu] in the range from 200 to 400 nm, using the respective solution as a blank. The  $\lambda_{max}$  of the API was 246.00 nm.

**Standard Curve of Atorvastatin calcium**

**Preparation of solution to calibration curve-** UV absorption shows lambda max to be 246.00 nm. The standard plot was prepared in methanol. From the standard stock solution (100µg/ml) take 0.5ml, 1.0ml, 1.5ml, 2.0ml, 2.5ml in 10ml volumetric flask & volume made upto the mark to obtained dilution of concentration 5µg/ml to 25µg/ml. Then absorbance was taken & calibration curve was plotted.

**Preformulation Studies of Atorvastatin calcium**

In physical characterization drug's physical state and solubility was measured.

**Melting Point Determination**

The melting point of a compound is the temperature at which it changes from a solid to liquid. Melting Point of the drug sample was determined by using melting point apparatus. A capillary tube was taken and one end was blocked by melting. A small amount of compound was placed on a clean surface. The compound was put into the open end of the capillary tube. The capillary tube was placed into the melting point apparatus. The sample was observed continuously for liquefaction and the melting point was recorded.

**Solubility study of Atorvastatin calcium:**

Solubility study was performed in Distilled water; 0.1N HCL, Methanol, Ethanol, pH 6.8 Phosphate buffer and Acetonitrile.

**Compatibility Screening of drug with Excipients**

For formulation development compatibility study between drug and excipients is necessary for stable dosage form on basis of its physical, chemical and biological characteristics. Drug-excipients interaction study was carried out by taking 1:1 ratio (w/w) of drug and excipients in 2ml glass vials, sealed and placed in stability chamber at 25°C/60% RH, 40°C /75 % RH and 60°C for 21 days. The sample was analysed for any changes in colour and odour after 7, 15, & 21 days.

**Infrared studies:**

**Characterization FTIR studies-** The IR studies were carried out by the pressed pellet technique using a KBr press in which the KBr was taken and kept in a hot air oven for two hours for the removal of any moisture. The above dried KBr was taken for the preparation of pellets of drug, and the selected formulations. The prepared pellet was placed in the sample holder and kept in the instrument to record the IR peaks. The results of the infrared studies for the selected batch are recorded.

**Preparation of batches Solid Dispersion**

Atorvastatin calcium solid dispersion can be formed by using the drug & polymer at different ratio in the present study solid dispersion of atorvastatin calcium can be formed by using poloxamer 188 by different ratio respectively.

**Table 2 Composition of batches solid dispersion**

Sr.No.	Formulation	Drug(mg)	Carrier(mg)
1	ATRP1	1	1
2	ATRP2	1	1.5
3	ATRP3	1	2
4	ATRP4	1	2.5
5	ATRP5	1	3
6	ATRP6	1	3.5

**Preparation of solid dispersion of Atorvastatin calcium (ATR)**

Solid dispersion of ATR in Poloxamer 188

was prepared in different ratios by solvent evaporation method. The Drug and carrier were dissolved in minimum volume of methanol &

solvent was removed under vacuum rotavapor at 40°C & 45rpm for 24hrs. The resultant solid dispersion was kept in refrigerator for 2 days to harden. Dispersion were then pulverized in mortar & pestle, passed through a 250-µm sieve (mesh size 60), then stored in a desiccator at room temperature.

#### Solubility studies of solid dispersion of ATR

Solubility may be defined as the amount of a substance that dissolves in a given volume of solvent at a specified temperature. Solubility measurements of Atorvastatin calcium were performed according to a published method by Higuchi & Connors. The amount of SD powder (mg) was weighed accurately in screw cap vials was dissolved in 5ml methanol by sonication for 15 minutes subsequently, the solutions were filtered through a Whatman filter paper no. 1. Filtered solution was diluted properly with methanol. The diluted solution was analyzed for the Atorvastatin calcium in UV at 246nm.

#### In vitro Dissolution study of solid dispersion

The dissolution rate of Atorvastatin calcium as such and from solid dispersions prepared was studied respectively in 900 ml of phosphate buffer pH 6.8 using USP type II (paddle type) dissolution test apparatus with a paddle stirrer at 75 rpm. A temperature 37±5°C was maintained throughout the study. Drug or solid dispersion equivalent to 40 mg of Atorvastatin calcium was used in each test. Samples of dissolution media (5ml) were withdrawn through a filter (0.45µ) at different intervals of time, suitably diluted and assayed at 246 nm. The samples of dissolution fluid withdrawn at each time were replaced with fresh fluid.

#### Drug content Analysis of solid dispersion

Solid dispersions equivalent to 10 mg of Atorvastatin calcium were weighed accurately & dissolved in the 10ml of methanol & volume was made up to 50 ml. From this 1ml of solution was taken and further diluted 10 times with methanol. The solution was filtered, diluted suitably and drug content was analyzed at 246 nm by UV spectrophotometer. The actual drug content was calculated using the following equation as follows:

$$\% \text{Drug content} = (M_{act}/M_t) \times 100 \quad (1)$$

$M_{act}$  = Actual amount of drug in Solid dispersion

$M_t$  = Theoretical amount of drug in solid dispersion

#### Evaluation of tablets

Physical Description: Tablets from all the batches & innovator were visually examined.

#### Diameter and Thickness

Diameter and thickness of tablets from all batches and innovator were measured by placing tablet between claws of digital calliper.

#### Average Weight and Weight Variation Test

Twenty tablets from each batch were weighed individually and the average weight was calculated. From average and individual weight, percentage weight variation was calculated.

#### Hardness Test

Hardness of tablet was determined using Monsanto hardness tester. The tester consists of a barrel containing a compressible spring held between two plungers. The lower plunger was placed in contact with the tablet and a zero reading was taken. The upper plunger was then forced against a spring by turning a threaded bolt until the tablet was fractured. As the spring was compressed, pointer ridded along a gauge in the barrel to indicate the force.

#### Friability Test

Friability was determined by taking thirty six tablets (pre-weighed) equivalent to 6.5g from each batch and placing them in a Roche Friabilator operated at 25 rpm for 4 minutes. Tablets were then dusted, reweighed and percentage friability was calculated.

$$\% \text{Friability} = (W_t \text{ initial} - W_t \text{ final} / W_t \text{ initial}) \times 100 \quad (2)$$

#### Disintegration Test

The disintegration apparatus consist of basket of six tubes with a base of metal sieve of 10 mesh. This assembly was suspended using a hanger with a mechanism of vertical motion at fixed speed of 28-32 cycles/minute in the phosphate buffer pH 6.8 maintained at 37±2°C. The time required for disintegration of tablet was recorded.

### Stability testing of API

The purpose of stability testing is to provide evidence on how the quality of a drug substance or drug product varies with time under the influence of a variety of environmental factors

such as temperature, humidity, and light, and to establish a re-test period for the drug substance or a shelf life for the drug product and recommended storage conditions.

**Table 3 Different storage conditions**

Study	Storage condition	Minimum time period covered by data at submission
Long term	25°C ± 2°C / 60% RH ± 5% RH	12 months
Intermediate	30°C ± 2°C / 65% RH ± 5% RH	6 months
Accelerated	40°C ± 2°C / 75% RH ± 5% RH	6 months

### III. RESULT & DISCUSSION

#### Method of analysis of Atorvastatin Calcium

The absorbance was measured in a UV

spectrophotometer at 246nm in Distilled water, pH 6.8 phosphate buffer, Methanol, Ethanol, 0.1N HCl and Acetonitrile. The scan observed in Methanol.

Concentration (µg/ml)	Absorbance			Average Absorbance
	1	2	3	
5	0.175	0.173	0.174	0.174
10	0.331	0.344	0.33	0.335
15	0.491	0.495	0.494	0.493
20	0.652	0.66	0.663	0.658
25	0.804	0.804	0.803	0.804
<b>Linear Equation</b>		y = 0.0322x + 0.0085		
<b>R<sup>2</sup> value</b>		R <sup>2</sup> = 0.9995		
<b>Slope (m)</b>		0.0322x		
<b>Intercept (c)</b>		0.0085		

**Table 4 Standard curve data of Atorvastatin Calcium in methanol**

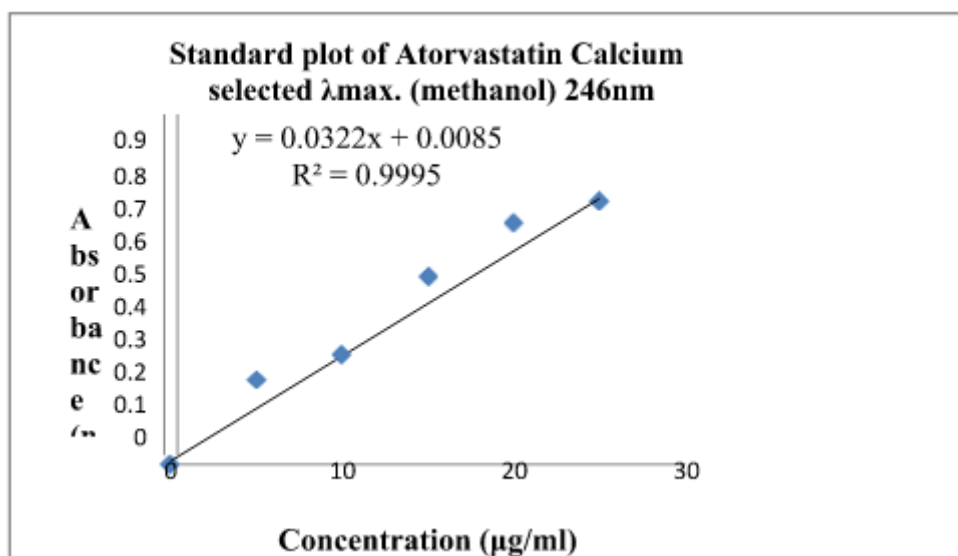


Fig 2: Standard Plot of Atorvastatin Calcium in Methanol

**Preformulation Studies of Atorvastatin calcium**

**Physical appearance:** A white to off-white, crystalline powder without any characteristic odour.

**Melting Point Determination**

The actual melting point of Atorvastatin calcium was determined by capillary method & it was found to be 157.7°C. This was in good agreement with the actual melting range.

**Compatibility Screening of drug with Excipients**

Various excipients were selected according to their function from the compatibility study report and based on review of literature and their concentration in formulation was optimized by taking different trial batches. No significant change was observed during compatibility studies of API and excipients at all the three conditions

**Table 5 Condition-25°C ±2°C /60%RH ± 5%RH**

S.No.	Ingredients	Initial	1 <sup>st</sup> Week	2 <sup>nd</sup> Week	3 <sup>rd</sup> Week
1	ATR	Whitepowder	NC	NC	NC
2	ATR+POLOXAMER 188	Whitepowder	NC	NC	NC
3	ATR+MCC	Whitepowder	NC	NC	NC
4	ATR+STARCH	Whitepowder	NC	NC	NC
5	ATR+SSG	Whitepowder	NC	NC	NC
6	ATR+TALC	Whitepowder	NC	NC	NC
7	ATR+Mag.Stearate	Whitepowder	NC	NC	NC

**Table 6 Condition-40°C ±2°C / 75% RH ± 5%RH**

S.No.	Ingredients	Initial	1 <sup>st</sup> Week	2 <sup>nd</sup> Week	3 <sup>rd</sup> Week
1	ATR	Whitepowder	NC	NC	NC
2	ATR+POLOXAMER 188	Whitepowder	NC	NC	NC



3	ATR+MCC	Whitepowder	NC	NC	NC
4	ATR+STARCH	Whitepowder	NC	NC	NC
5	ATR+SSG	Whitepowder	NC	NC	NC
6	ATR+TALC	Whitepowder	NC	NC	NC
7	ATR+Mag. Stearate	Whitepowder	NC	NC	NC

Table7 Condition-60°C ±2°C

S.No.	Ingredients	Initial	1 <sup>st</sup> Week	2 <sup>nd</sup> Week	3 <sup>rd</sup> Week
1	ATR	Whitepowder	NC	NC	NC
2	ATR+POLOXAMER188	Whitepowder	NC	NC	NC
3	ATR+MCC	Whitepowder	NC	NC	NC
4	ATR+STARCH	Whitepowder	NC	NC	NC
5	ATR+SSG	Whitepowder	NC	NC	NC
6	ATR+TALC	Whitepowder	NC	NC	NC
7	ATR+Mag. Stearate	Whitepowder	NC	NC	NC

Where ATR (Atorvastatin calcium IP), \*\* (same as initial), NC (No change)

**Results of FT-IR spectrum interpretation**

The IR spectrum of ATR is shown in figure 6.4.

The IR spectrum showed various characteristic peaks. All observed peaks justifies the presence of all functional groups present ATR.

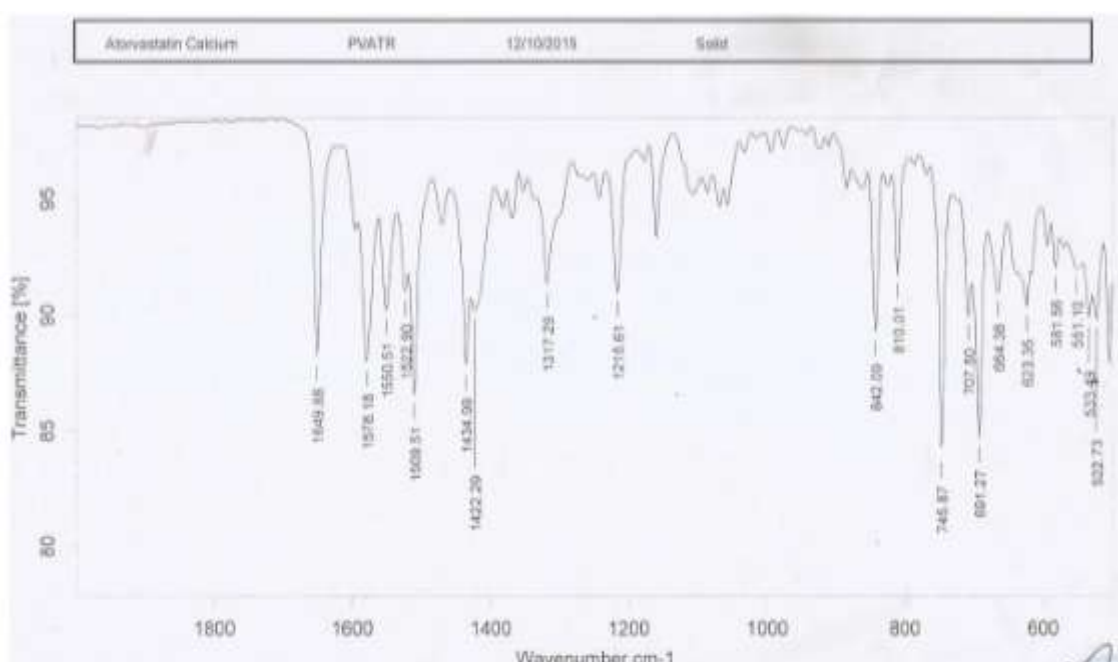


Fig. 3 FT-IR spectra of Atorvastatin calcium

**Table 8 FT-IR peaks and functional category**

Peaks( $\text{cm}^{-1}$ )	Functional groups
1435	-CH <sub>2</sub> bending
1216	C-O stretching
1650	C=O stretching
1578	N-H bending
1510	C=C stretching in aromatic
1317	C-N stretching

The infrared spectrophotometer study was carried out for the API and optimized batch for extended release tablet. The IR data and IR spectrum for the API excipients and tablet are indicated in Figure 6.8 respectively. The FTIR data for the API indicated that the functional group's peaks indicated in spectra are identical with the theoretical peaks, so this indicates the purity of the API. The peaks of all the functional groups in FT-IR spectra of optimized formulation are identical with that of the pure drug XYZ. From the IR studies it was found that there was no interaction of

the drug-excipient and excipient-excipient because there is no change in functional groups peak. So, all the excipients are compatible with API.

**Solubility study of solid dispersion of Atorvastatin calcium**

Solubility of all solid dispersion can be increased as compared to pure drug by solid dispersion increasing the concentration of polymer. The formulation ATRP4 showed the more solubility than pure drug (ATR).

Sr. No.	Formulation	Drug	Solubility $\mu\text{g/ml}$
1	ATR	Pure drug	2.05
2	ATRP1	1:1	9.16
3	ATRP2	1:1.5	13.22
4	ATRP3	1:2	19.34
5	ATRP4	1:2.5	23.09
6	ATRP5	1:3	16.24
7	ATRP6	1:3.5	12.19

Table 9 Solubility study of solid dispersion



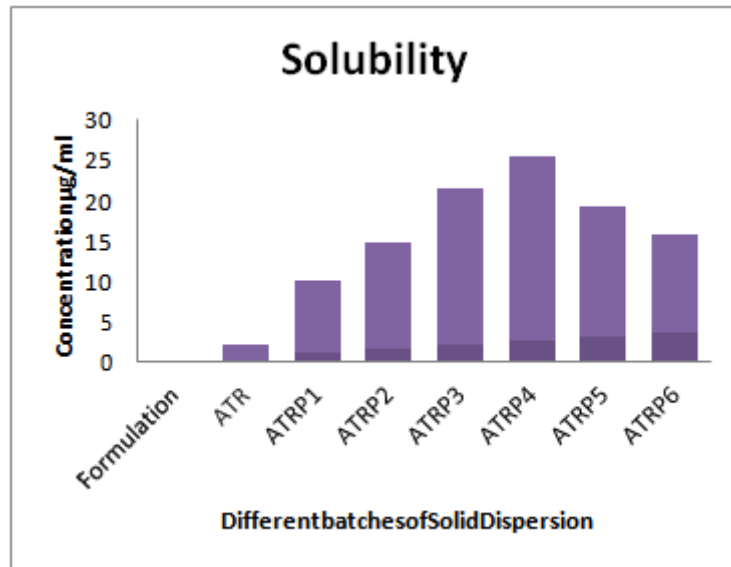


Fig 4 Solubility study of solid dispersion.

**In-vitro Dissolution Study of Solid Dispersion**

The dissolution profile of the solid dispersion was shown in figure 6.6. The dissolution rate of

Atorvastatin calcium solid dispersion was higher for all the formulation when compared with pure drug (ATR).

**Table 10 In-vitro dissolution data of solid dispersion**

Time (min)	Innovator product (Atorvastatin 40 mg)	ATRP1	ATRP2	ATRP3	ATRP4	ATRP5	ATRP6
00.00	00.00	00.00	00.00	00.00	00.00	00.00	00.00
5	25.32	24.77	25.08	25.10	27.6	26.9	27.4
10	43.71	41.1	40.2	39.89	41.6	41.26	39.6
15	61.20	60.11	61.3	59.36	60.4	61.2	59.20
20	73.80	74.3	72.24	75.45	74.7	72.7	74.2
25	86.90	82.25	83.5	82.6	85.3	82.55	81.5
30	95.65	91.35	92.83	91.10	93.58	91.24	91.5

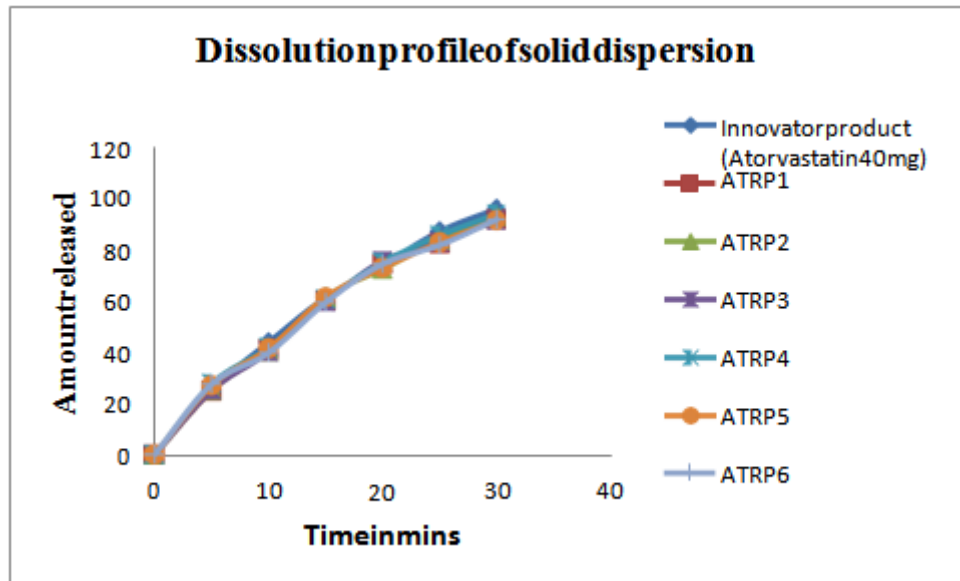


Fig5 Dissolution study of solid dispersion

Drug Content Analysis

Table 11 Drug content analysis of solid dispersion

S.no	Formulation code	Drug: Carrier	% Drug Content
1	ATRP1	1:1	91.27
2	ATRP2	1:1.5	92.34
3	ATRP3	1:2	91.99
4	ATRP4	1:2.5	94.28
5	ATRP5	1:3	93.84
6	ATRP6	1:3.5	90.21

Formulation of Tablet from Solid Dispersion

Table 12 Formula for tablet (mg)

Ingredients	ATRP1	ATRP2	ATRP3	ATRP4	ATRP5	ATRP6
Atorvastatin CaIP	40	40	40	40	40	40
MCCIP	49	47	45	51	53	55
StarchIP	62	64	66	60	58	56
SSGIP	25	25	25	25	25	25
Mg stearateIP	1.5	1.5	1.5	1.5	1.5	1.5
TalcumIP	2.5	2.5	2.5	2.5	2.5	2.5
Total weight (mg)	180	180	180	180	180	180

Where- SSG is sodium starch glycolate and MCC is microcrystalline cellulose.

**Rheological properties**

**Table 13 Flow Properties of granules**

Parameters	ATRP1	ATRP2	ATRP3	ATRP4	ATRP5	ATRP6
Angle of repose (degrees)	34.29	32.76	32.48	<b>32.68</b>	31.21	32.41
Bulk density (g/ml)	0.563	0.612	0.569	<b>0.626</b>	0.591	0.643
Tapped density (g/ml)	0.746	0.820	0.754	<b>0.830</b>	0.766	0.846
% Compressibility index	24.53	25.37	24.54	<b>24.58</b>	22.85	23.40
Hausner ratio	1.33	1.34	1.32	<b>1.33</b>	1.30	1.32

**Evaluation of tablets:**

All the formulations showed values within the prescribed limits for tests like hardness, friability, weight variation, disintegration time and

drug release which indicate that the prepared tablets are of standard quality. Here Innovator product is Atorvastatin 40 mg is used.

**Table 14 Evaluation of tablet of Atorvastatin calcium solid dispersion**

Batch	ATRP1	ATRP2	ATRP3	ATRP4	ATRP5	ATRP6	Innovator
Diameter (mm) (8.0 ± 0.2)	8.01±0.06	7.98±0.09	8.00±0.06	7.96±0.05	7.98±0.06	7.99±0.09	8.00±0.10
Thickness (mm) (3.5 ± 0.3)	3.27±0.03	3.30±0.03	3.27±0.01	3.29±0.06	3.28±0.03	3.28±0.01	3.28±0.04
Average weight (mg)	180.42	180.15	180.74	180.51	181.14	181.02	180.12
% Weight variation (within ±7.5% of Av. Wt.)	-3.16 to +4.12	-2.58 to +3.41	-3.15 to +4.34	-2.48 to +2.05	-3.15 to +4.34	-4.19 to +3.40	-2.01 to +2.00
Hardness (kg/cm <sup>2</sup> )	2.63±0.17	3.68±0.21	3.53±0.13	3.25±0.07	4.12±0.17	3.69±0.06	3.26±0.24
Friability (%) (NMT 1.0%)	0.31	0.16	0.10	0.15	0.04	0.16	0.09
Disintegration time (min) (NMT 15 min)	6.25	13.40	12.21	7.15	14.50	9:34	7.13

The shape of the tablets of all formulations remained circular & biconvex with no visible cracks, capping or lamination. The disintegration time of all batches remained within the range of 6.25minutes to 14.50minutes which is within Pharmacopoeial limit. The diameter, thickness, average percentage weight variation,

hardness, percentage friability, disintegration time and drug release within 30minutes of innovator product (Atorvastatin 40mg) were found to be  $8.00 \pm 0.10$ mm,  $3.28 \pm 0.04$ mm, within  $\pm 5\%$ ,  $3.26 \pm 0.24$ kg/cm<sup>2</sup>, 0.09%, 7:13minutes and 95.65% respectively.

**In-Vitro Dissolution Studies of tablets**

**Table 15 In-vitro dissolution data (cumulative %drug release) of tablets**

Time(mi n.)	Innovatorproduct(Atorvastatin 40mg)	ATRP1	ATRP2	ATRP3	ATRP4	ATRP5	ATRP6
<b>00.00</b>	<b>00.00</b>	<b>00.00</b>	<b>00.00</b>	<b>00.00</b>	<b>00.00</b>	<b>00.00</b>	<b>00.00</b>
<b>5</b>	<b>25.32</b>	26.77	25.12	26.30	<b>27.6</b>	26.9	27.4
<b>10</b>	<b>43.71</b>	42.1	40.4	40.05	<b>41.6</b>	41.26	44.6
<b>15</b>	<b>61.20</b>	60.11	61.3	59.36	<b>60.4</b>	61.2	60.21
<b>20</b>	<b>73.80</b>	75.3	77.3	76.45	<b>74.7</b>	75.7	74.2
<b>25</b>	<b>86.90</b>	85.25	86.6	87.6	<b>85.3</b>	85.55	84.5
<b>30</b>	<b>95.65</b>	92.49	93.84	92.10	<b>94.26</b>	91.62	92.3

The ATRP4 was optimized on the basis of resemblance with innovator (Atorvastatin 40mg) in terms of better drug release profile, disintegration time, hardness, dissolution and friability of drug as compared to other formulation batches. The result was found to be satisfactory as innovator product (Atorvastatin 40mg).

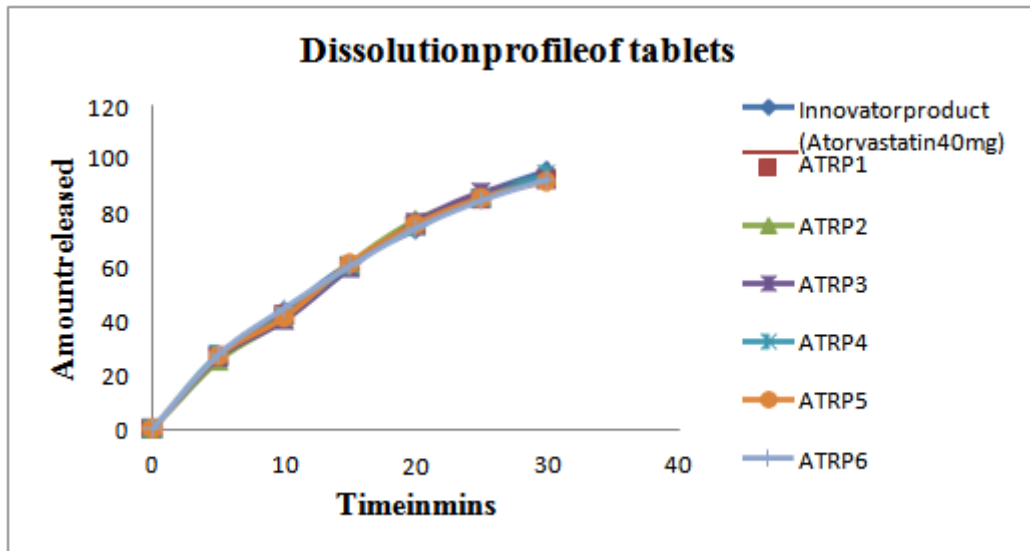


Fig 6 In-Vitro Dissolution Studies from tablets

Table 16 In-vitro dissolution data of ATRP 4 & innovator

Dissolution time (minutes)	Cumulative Drug release (%)	
	ATRP4	Innovator (Atorvastatin 40mg)
0	0	0
5	27.6	25.32
10	41.6	43.71
15	60.4	61.20
20	74.7	73.80
25	85.3	86.90
30	94.26	95.65

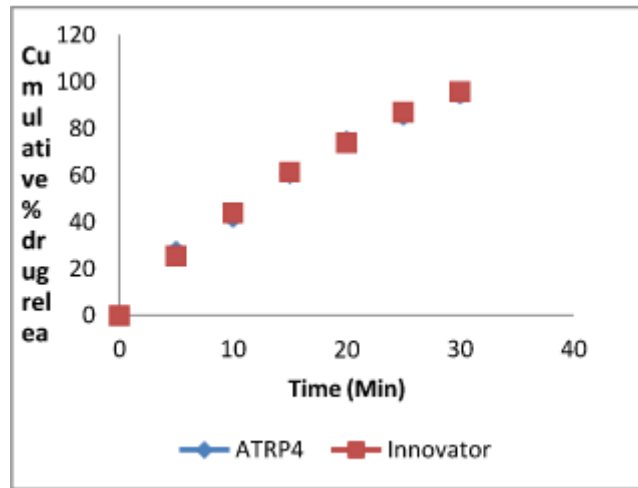


Fig. 7 Comparative dissolution profile of SD tablet of ATRP4 & innovator

**Drug Release kinetic**

The in-vitro release data obtained were fitted in to various kinetic equations (i.e. zero, first, higuchi, korsmeyer, hixson and peppas kinetic

model) of the optimized formulation ATRP4 was following the higuchi more linearly than other models.

Table 17 Results of drug release kinetics using various models

Formulation Code	R <sup>2</sup> Values				n-Value
	Zeroorder	firstorder	Higuchimodel	Korsmayer-peppas	
ATRP4	0.9641	0.9562	0.9817	0.9599	0.9927

**Stability testing of Atorvastatin calcium (Optimized formulation) :** The purpose of stability testing is to provide evidence on how the quality of a drug substance or drug product varies with time under the influence of a variety of environmental factors such as temperature, humidity, and light, and to establish a re-test period for the drug substance or a shelf life for the drug product and recommended storage conditions.

In general, a drug substance should be

evaluated under storage conditions (with appropriate tolerances) that test its thermo stability and, if applicable, its sensitivity to moisture re-test period for the drug substance or a shelf life for the drug product and recommended storage conditions. In general, a drug substance should be evaluated under storage conditions (with appropriate tolerances) that test its thermo stability and, if applicable, its sensitivity to moisture

Table 18 Stability data of optimized batch ATRP4 at 30 0 C ± 2 0 C/65% RH ± 5%RH

Test protocol	Specification	Sampling time (in months)			
		Initial	1 M	2 M	3 M
Description	Circular, biconvex	Complies	Complies	Complies	Complies



Disintegration time(min.)	NMT30	7:31	6:50	7:20	7:18
% Drug release (in 30 min.)	NLT75% in 45 min.	93.81	94.26	92.67	91.79
Assay: Atorvastatin Caq. to Atorvastatin 40 mg	38-42mg/tablet 90-110%	39.86 mg 99.30%	39.80 mg 99.00%	39.72 mg 98.60%	39.60 mg 98.00%

**Table 19 Stability data of optimized batch ATRP4 at 40 °C ± 2 °C/75% RH ± 5% RH**

Test protocol	Specification	Sampling time (in month)			
		Initial	1 M	2 M	3 M
Description	Circular, biconvex	Complies	Complies	Complies	Complies
Disintegration time (min.)	NMT30	7:31	7:27	7:53	7:49
% Drug release (in 30 min.)	NLT85% in 45 min.	93.81	92.26	90.94	89.79
Assay: Atorvastatin Caq. to Atorvastatin 40mg	38-42mg/tablet 90-110%	39.86 mg 99.30%	39.77 mg 98.85%	39.61 mg 98.05%	39.43 mg 97.15%

**SHELF-LIFE AND ITS CALCULATION**

**Table 20 Shelf-life of optimized formulation ATRP4**

S.No.	Parameters	Storage conditions	
		30°C/65%RH	40°C/75%RH
1	K(days <sup>-1</sup> )	1.54 x 10 <sup>-4</sup>	2.303 x 10 <sup>-4</sup>
2	t <sub>1/2</sub> (days)	4500	3009.12
3	t <sub>0.9</sub> (days)	681.82	455.93

Solid dispersion of Atorvastatin calcium tablets stored at  $30 \pm 2^\circ\text{C}/65\% \text{RH} \pm 5\% \text{RH}$  showed K value as  $1.54 \times 10^{-4} \text{ day}^{-1}$ ,  $t_{0.9}$  value as 681.82 days and  $t_{1/2}$  value as 4500 days, while those stored at  $40 \pm 2^\circ\text{C}/75\% \text{RH} \pm 5\% \text{RH}$  showed K value as  $2.303 \times 10^{-4} \text{ day}^{-1}$ ,  $t_{0.9}$  value as 455.93 days and  $t_{1/2}$  value as 3009.12 days.

The value of K obtained in case of Atorvastatin calcium tablets stored at  $40^\circ\text{C} \pm 2^\circ\text{C}/75\% \text{RH} \pm 5\% \text{RH}$  was found to be greater than those stored at  $30^\circ\text{C} \pm 2^\circ\text{C}/65\% \text{RH} \pm 5\% \text{RH}$ .

The results of stability studies suggest that for adequate shelf life of Atorvastatin calcium tablet, the ideal storage temperature is not to exceed  $30^\circ\text{C}$  and relative humidity not to exceed 65%.

#### IV. CONCLUSION

Atorvastatin calcium was identified & characterized as per requirements of official monograph. Melting point range was found to be in between  $157.7^\circ\text{C}$ , which is near to the official range  $159.2-160.7^\circ\text{C}$  as per literature. Atorvastatin calcium solid dispersions were prepared by solvent evaporation method. Solubility of all solid dispersion can be increased as compared to pure drug by solid dispersion increasing the concentration of polymer. The formulation ATP4 ( $23.09 \mu\text{g/ml}$ ) showed the more solubility than pure drug ( $2.05 \mu\text{g/ml}$ ). Drug content of all solid dispersion is acceptable range. The dissolution rate of Atorvastatin calcium solid dispersion was higher for all the formulation when compared with pure drug. It was concluded that there was no significant change in the formulated tablets even after solid dispersion. Thus the solid dispersion of Atorvastatin calcium tablets, comparable with the innovator formulation (Atorvastatin 40mg), was successfully developed which could overcome the problem of low solubility. Tablets of optimized ATP4 batch were kept for stability study at  $30 \pm 2^\circ\text{C}/65\% \text{RH} \pm 5\% \text{RH}$  and  $40 \pm 2^\circ\text{C}/75\% \text{RH} \pm 5\% \text{RH}$  and it was found that the drug degradation rate constant at  $40 \pm 2^\circ\text{C}/75\% \text{RH} \pm 5\% \text{RH}$  was greater than those stored at  $30 \pm 2^\circ\text{C}/65\% \text{RH} \pm 5\% \text{RH}$ . Finally it was concluded that for better stability Atorvastatin calcium tablet should be stored at a temperature and RH not exceeding  $30^\circ\text{C}$  and 65% respectively.

#### REFERENCES

[1]. Adams, L.B., Hyperlipidemia., Guidelines for Adolescent Nutrition Services, 2005,

[2]. Tripathi K.D., "Essentials of Medical pharmacology", 6 th Edition, Jaypee Brothers Medical Publisher, New Delhi, 2006, 612-616.

[3]. Rohilla Ankur and Dagar Nidhi, et al, "Hyperlipidemia: A Deadly Pathological Condition" International Journal Of Current Pharmaceutical Research, 2012, vol 4, 15-18.

[4]. Sharma Monika and Garg Rajeev, et al, "Formulation and evaluation of solid dispersion of Atorvastatin Calcium" Journal of pharmaceutical and Scientific Innovation, 2012, 73-81.

[5]. Brahmanekar M.D and Sunil B.J, "Biopharmaceutics and Pharmacokinetics a Treatise", 2 nd edition, Vallabh Prakashan, 2009, 335-336.

[6]. Kumar K Hemanth Pavan & Chandramouli Yerram, et al, "Enhancement Of Solubility & Dissolution Rate Of Diclofenac Sodium By Solid Dispersion" International Journal Of Advanced Pharmaceutics, 2012, vol-2, 110-118.

[7]. T. Ketan Savjani & K. Gajjar Anuradha et al, "Drug solubility improvement & enhancement techniques" International Scholarly Research Network, 2012 vol-1, 1-10

[8]. K.M vidhya & T.R Saranya, et al, "Pharmaceutical Solid Dispersion Technology: A Promising Tool to Enhance Oral Bioavailability" International Research Journal Of Pharmaceutical & Applied Science, 2013, 3(5), 214-218.

[9]. Zameeruddin Mohmed & Rajmalle Kishor R., et al, 2014 "Recent approaches solubility & dissolution enhancement of atorvastatin " World Journal Of Pharmacy & Pharmaceutical Sciences, vol-3, 534-544.

[10]. Sarkar R, Sultana R et al, "Improvement solubility of atorvastatin calcium using solid dispersion techniques" International journal of pharmaceutical sciences & research, 2014, vol 5 (12), 5405-5410.

[11]. Ugandhar, C, et al, "Design, development and evaluation of immediate release drug combination", Asian Journal of Pharmaceutical Clinical and Research, 2011, Vol. 4, Issue 3, 77-79

[12]. Patel B Bhavesh, & Patel K



- Jayvadan, et al, “ revealing facts behinds spray dried solid dispersion technology used foe solubility enhancement ”, Saudi Pharmaceutical Journal, 23, 2015, 352-365
- [13]. Ting Yuan & Lingzhen Qin, et al, “solid lipid dispersion of calcitriol with enhanced dissolution & stability”, Asian Journal Of Pharmaceutical Sciences, 2015, 08, 352-365
- [14]. ICH harmonized tripartite guideline, Stability Testing Of New Drug Substances And Products (R2), 1-15.
- [15]. Huang Yanbin and Dai Wei-guo, “ Fundamental Aspects of Solid Dispersion Technology for Poorly Soluble Drug” Acta Pharmaceutica Sinica B, 2014, 18-25.