

## Biological Activities of Imidazole Derivatives: A Review

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### ABSTRACT:

Imidazole is a five-membered, planar heterocyclic ring with 3C, 2N, and N in the first and third positions. Purine, histamine, histidine, and nucleic acids are just a few examples of important natural compounds containing the imidazole ring. Because it is an aromatic chemical that is polar and ionizable, it is used as a treatment to improve the solubility and bioavailability properties of proposed weakly soluble chemical entities and thus improves basic pharmacokinetic parameters of lead molecules. Imidazole derivatives hold a special place in medicinal chemistry. The introduction of the imidazole nucleus is a significant synthesis technique in the method used to find pharmaceutical medications. This article aims to review previous years' work on imidazole chemistry and biological activities.

**KEYWORDS:** Imidazole, antibacterial, antifungal, heterocyclic, biologically active.

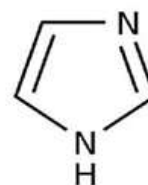
### I. INTRODUCTION

Imidazole is a five-member heterocyclic aromatic compound with two Nitrogen atoms that are sp<sup>2</sup> hybridized. Because the imidazole ring contains two types of lone pairs, delocalized and non-delocalized (non-Huckle-lone pair), the pKa of each Nitrogen differs. Nitrogen with a delocalized lone pair has pKa=7, while nitrogen with a non-delocalized lone pair has pKa=14.9. As a result, Imidazole is in amphoteric nature, which means it can act as both an acid and a base and is susceptible to nucleophilic and electrophilic attack [1]. Imidazole is a colorless or pale-yellow solid with an amine-like order. It is an aromatic heterocyclic categorized as a diazole and an alkaloid. It dissolves in water and other polar solvents. Because the hydrogen atom can be found on either of the two nitrogen atoms, it exists in two equivalent tautomeric forms. The melting point of imidazole is 88.9°C and the boiling point is 267.8°C. Imidazole is polar in nature and its dipole

moment is 4.8 Debye, the molecular formula is C<sub>3</sub>H<sub>4</sub>N<sub>2</sub>, and the structural formula [2,3].

Imidazole is a heterocyclic class with a five-membered ring structure and variable substituents. This ring system is found in important biological skeletons such as histidine and the associated hormone histamine. Imidazole can act as both a weak acid and a base. Nitroimidazole and antifungal drugs are examples of drugs with an imidazole ring. [4]

Heterocyclic compounds are useful in both pharmacology and agriculture. An examination of research manuscripts from the past 10 decades revealed a general pattern of research for novel pharmaceuticals involving modifications to current physiologically robust matrices and molecular approaches of the compounds' structures.



In drug discovery, the imidazole nucleus is an important synthetic technique. Imidazole derivatives have anti-inflammatory, anti-cancer, antimicrobial, analgesic, and antitubercular properties. [5,6] One of the most important properties of imidazole derivatives is their use as a material for the treatment of denture stomatitis. The high beneficial properties of imidazole-associated drugs have encouraged medicinal chemists to prepare a large number of new chemotherapeutic materials. Imidazole drugs have a wide scope in the pharmaceutical field. [7]

### PHARMACOLOGICAL ACTIVITIES:

Imidazoles are well-known heterocyclic compounds that are common and have an important feature in a variety of medicinal agents. [8] Based on various literature surveys, imidazole

derivatives show various pharmacological activities:

1. Antifungal activity
2. Anticancer activity
3. Antibacterial activity
4. Anti-tubercular activity
5. Anti-HIV activity
6. Anti-inflammatory and analgesic activity
7. Antiviral activity
8. Anthelmintic activity
9. Antidepressant activity

### 1) ANTI-FUNGAL ACTIVITIES:

In recent years, imidazole and triazole chemists have been the main areas of attention in the hunt for novel antifungals. Unquestionably, theazole family of medications, a variety of 1-substituted imidazole and triazole chemicals, constitutes the current method for treating fungal illness both topically and systemically. [9] Imidazole has strong pharmacological and biochemical actions as an anti-fungal. Due to poor absorption and substantial first-pass metabolism, the lipophilic Imidazoles, including clotrimazole, econazole, and miconazole, showed poor systemic availability after oral administration. As a result, their usage has been restricted to the topical treatment of superficial fungal infections. Ketoconazole and more popular imidazole introduced into therapy in the late 1970s represented a breakthrough in the treatment of antifungal disease. [10]

### 2) ANTI-CANCER ACTIVITY:

To test their anticancer properties, several new imidazoles-(Benz) azoles and imidazole piperazine derivatives were synthesized. [11] According to anticancer activity screening findings, these compounds were the most potent in the group.

### 3) ANTIBACTERIAL ACTIVITY:

According to the literature study, the antibacterial action of imidazole derivatives is the second most frequent significant pharmacological effect. Finding this impact is significant because, with the discovery of nearly all major antibiotic groups (tetracyclines, cephalosporins, aminoglycosides, and macrolides), these medications may become less effective due to the rise in microbial resistance. Currently, multidrug-resistant bacteria-related treatment failures are a major public health concern on a global scale. [12]

For instance, researchers looked into the bactericidal effects of imidazole compounds in combination with silver. The National University of Ireland's John McGinley et al. (2013) synthesized 1-(3-

amino propyl) imidazole and produced Schiff base ligands that were simple to couple with Ag(I) centers. Studies were conducted on *S. aureus*, MRSA, *E. coli*, and *P. aeruginosa* strains. The majority of Ag (I) complexes exhibited moderate antibacterial activity as a consequence. [13]

The antibacterial research of benzimidazole hexafluorophosphate and coumarin salt substituted with imidazolium, benzimidazolium, and silver complexes against Gram-positive (*S. Aureus*) and Gram-negative (*E. coli*) bacteria was carried out by (Jawaharlal Nehru Centre, India). While the antibacterial activity against *S. aureus* was only mild, both series of silver complexes demonstrated antibacterial action against *E. coli*. The complicated action is connected to the metal center, it was finally determined. [14]

### 4) ANTI-TUBERCULAR ACTIVITY:

Despite recent advancements in the treatment of infectious illnesses brought on by Mycobacterium, these germs continue to pose a serious threat to global healthcare and are the main cause of infectious disease-related fatalities worldwide. Despite the existence of anti-tuberculosis medications, TB remains one of the most prevalent illnesses that warrant global attention. The HIV epidemic has aggravated the situation by increasing the frequency of multidrug-resistant TB and the development of drug-resistant microorganisms. [15] Finding novel treatment drugs to fight M. tuberculosis infections is necessary for light of these findings.

Researchers from the University of Pardubice in the Czech Republic, Daniel Cvejn, Vera Klimesova, and Filip Bures, examined the antimycobacterial properties of 2-phenyl imidazole derivatives made from amino acids in 2012. Nitro group-containing molecules among 2-phenyl imidazole derivatives exhibited the ability to inhibit M. tuberculosis, however this ability was less potent than that of isoniazid. Isoniazid's effectiveness was surpassed by *M. avium* and *M. kansasii* activity. The primary factor influencing the antimycobacterial action of the compounds investigated was the availability of the nitro group.

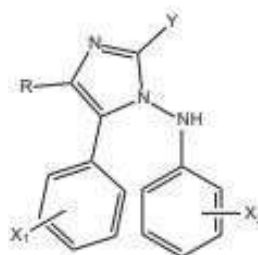
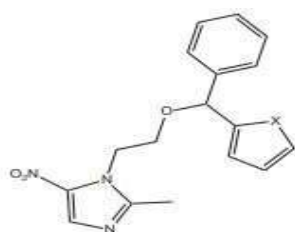
### 5) ANTI-HIV ACTIVITY:

AIDS or the AIDS Related Complex (ARC) is caused by the pathogenic retrovirus known as HIV-1 (Human Immunodeficiency Virus Type-

1). HIV infection causes severe deficiencies in cell-mediated immunity and targeting the monocytes that express surface CD4 receptors. Opportunistic infections (OIs), such as bacterial, fungal, viral, protozoal, and neoplastic disorders, as well as eventual mortality, are caused by opportunistic infections (OIs) over time due to substantial depletion of CD4 T-lymphocytes (T-cells) caused by infection [17]. An ideal anti-HIV drug would be able to fight off opportunistic illnesses including hepatitis, TB, and other bacterial infections in

addition to inhibiting HIV reproduction.

Imidazoles have a history of being used as antiviral medications; capravirine [18] is one such instance. Several 1-2-(diarylmethoxy)ethyl] were synthesised by Silvestri et al. [19] and De Martino et al. [20, 21]. The racemic 1-2-[(thiophen-2-yl)phenylmethoxy]ethyl]-2-methyl-5-nitroimidazole (EC<sub>50</sub>, 0.03 mol<sup>-1</sup>) being the most potent among all the analogues (Fig. 1), exhibiting higher activity than efavirenz against the virus IRT carrying the K103N mutation.

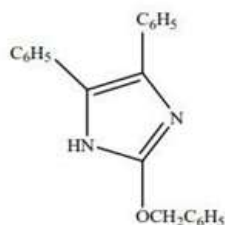


(DAMNIs X: S, O NAIMs X, X: halogen, alkyl R: alkyl, aryl Y: S, H, S)

Similarly, N-aminoimidazoles (NAIMs) have also been reported to inhibit replication of the WT virus as well as an HIV-1 strain that contained both the K103N and Y181C mutations.

#### 6) ANTI-INFLAMMATORY AND ANALGESIC ACTIVITY:

Using the Carrageenan-induced paw edema technique, a research on 2-substituted-4, 5-diphenyl-1H-imidazoles examined the anti-inflammatory effects. When compared to indomethacin, this chemical has the highest level of effect.



2-(benzyloxy)-4,5-diphenyl-1H-imidazole

#### 7) ANTIVIRAL ACTIVITY:

These are synthetic imidazole derivatives that have been tested for antiviral activity as 2-(substituted phenyl) imidazol-1-yl. Compounds A and B were shown to be the most

effective antiviral medicines when methanone was used against virus strains.

#### 8) ANTHELMINTIC ACTIVITY:

It was shown that extra-intestinal parasites, notably intravascular and intestinal lysing parasites, are less vulnerable to imidazole than gastrointestinal parasites. In comparable settings, the activity against developing phases is superior to that against arrested or adult stages. At levels that are ineffective in preventing the formation of an adult in vivo, larval development and hatching are hindered. They must be effective against nematodes at levels lower than those used to control cestodes and trematodes. [14] A greater dosage of medication or numerous treatments are required for cestode or trematode management. It has been discovered that the class member (2-alkylbenzimidazole) may eliminate several nematode and trematode species from diverse hosts. Tetra chloro-2-trifluoromethylbenzimidazole (4, 5, 6, 7) exhibits strong action against the nematodes *Fasciola hepatica*, *Ancylostoma caninum*, *Haemonchus contortus*, and *ascaris*. It has been discovered that many 2-5 disubstituted benzimidazoles, which have the ability to kill a number of different species of intestinal nematodes, also show action against cestodiasis in humans and animals.

Mebendazole 100mg/kg is used to treat patients with *T. solium* and *T. saginata*.

#### 9) ANTIDEPRESSANT ACTIVITY:

Moclobemide analogues were created by substituting substituted imidazole for the moclobemide phenyl ring, and their ability to treat depression was tested using the forced swimming method. It was discovered that analogue 7a-c was more effective than moclobemide.

### II. CONCLUSION:

This review on various imidazole derivatives, a significant class of heterocyclic compounds, has fascinating results for its antibacterial, anticancer, antitubercular, antifungal, analgesic, and anti-HIV activities. It also showed promising results for most pharmacological activities. Modifications to the imidazole nucleus have so far been seen to have interesting biological activity. It would be fascinating to see how many additional pharmacological profiles are added to it in the future because they are still unknown and may be used as a guide for future research to produce a few more potent molecules.

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