

Synthesis, Spectral analysis and Antimicrobial Activity of Some Newly Derivatives of 5 – amino Salicylic Acid.

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ABSTRACT

In the present investigation Synthesis, Spectral analysis and antimicrobial studies on some newly synthesized 5 -amino salicylic acid derivatives. All the newly derivatives were Synthes- sized by reflux method. The resulting crude precipitates were recrystallized from the proper solvent. The compound was analysed using CHNS Analysis, IR spectra, ¹H and ¹³C NMR spectroscopic. These derivatives have been screened for their antimicrobial activities and antifungal.

Key words: Synthesis, Spectral analysis, Antimicrobial and 5-ASA Derivatives.

I. INTRODUCTION

5-Amino salicylic acid also known as mesalamine or mesalamine, is an anti-inflammatory drug used to treat inflammation of the digestive tract Ulcerative colitis [1] and mild - to moderate Crohn's disease [2]. It is also recommended therapy for the induction and maintenance of remission of ulcerative colitis (UC) [3-4]. The drug acts topically at the colonic mucosa to reduce mucosal inflammation [5] yet because the active drug is rapidly absorbed in the stomach and small intestine [6] a number of oral formulations have been developed to deliver 5- amino salicylic acid to the colon [5,7]. The most common side effect of 5- amino salicylic acid are headache and flatulence. Hair loss and itching also may occur. Infrequent side effects include increased heart rate, Pancreatitis back pain fatigue, tremor. and ear pain and blood disorders.

The most important bimolecular, now a day with drastically different properties is required for various applications Chelates of biologically important molecules are also being investigated for various requirements of human life. Organic

molecules with donor atoms like N, O etc. are very good examples that can form coordination compounds. They show important biological and chemical properties. The derivatives of 5 - amino salicylic acid is used of medicinal purpose. Practically only few scientists have made attempt to study with 5-amino salicylic acid derivatives or biochemical formation and catalytic behaviours of 5 -amino salicylic acid derivatives.

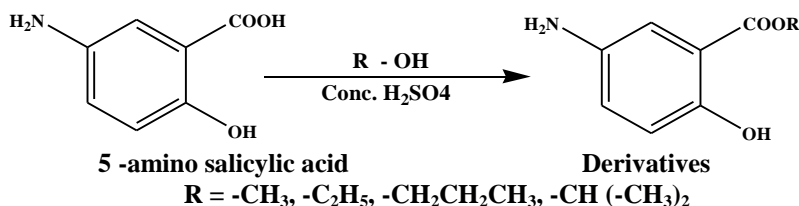
Looking to the literature survey carried out as well as the significance of the 5-amino salicylic acid derivatives as well as its coordination compounds, it is quite likely to give modified and improvised biochemical. Prompted by the above biological properties of 5- amino salicylic acid, it was contemplated to synthesize a newly series of 5 - amino salicylic acid derivatives. Antibacterial activities of the newly synthesized derivatives are discussed in this paper. The derivative of 5- amino salicylic acid contains - COOH, -OH (Phenolic) and -NH₂ groups.

EXPERIMENTAL

All chemicals used were of A.R grade and used & as such without further purification except for ethanol. 5- amino salicylic acid was obtained from s-d fine chemical company. Melting point were determined in open capillary tubes and are uncorrected, IR spectra (4000-400 cm⁻¹) were recorded on Shimadzu Pekin Elmer 8201 ET-IR with KBr pellets. The ¹H-NMR spectra and ¹³C NMR spectra were recorded on BRUKER AVANCE II 400 MHz Spectrometer.

Chemical shift values are reported as values in ppm relative to TMS (δ = 0) as internal standard in CDCl₃ solvent. Elemental analysis was performed on Vario MICRO C, H, N, S Elemental Analyzer system.

Synthesis of Derivatives:



(A) Methyl Derivatives:

A solution of 5 - amino salicylic acid (10.0 gm, 65.3 mmol) and sulfuric acid in methanol was heated under reflux in water bath. After addition of NaHCO₃ (until the evolution of CO₂ gas) the reaction mixture was filtrate. The filtrate was poured into water and extracted with ether. The combine organic layers were dried over Magnesium sulphate and the solvent was removed [8-9].

(B) Ethyl Derivatives:

A solution of 5 - amino salicylic acid (10.0 gm, 65.3 mmol) and sulfuric acid in ethanol was heated under reflux in water bath. After addition of NaHCO₃ (until the evolution of CO₂ gas) the reaction mixture was filtrate. The filtrate was poured into water and extracted with ether. The combine organic layers were dried over Magnesium sulphate and the solvent was removed [8-9].

(C) Propyl Derivatives:

A solution of 5 - amino salicylic acid (10.0 gm, 65.3 mmol) and sulfuric acid in n-propyl alcohol was heated under reflux in water bath. After addition of NaHCO₃ (until the evolution of CO₂ gas) the reaction mixture was filtrate. The filtrate was poured into water and extracted with petroleum ether. The combine organic layers were dried over Magnesium sulphate and the solvent was removed [8-9].

(D) Iso propyl Derivatives:

A solution of 5 - amino salicylic acid (10.0 gm, 65.3 mmol) and sulfuric acid in iso propyl alcohol was heated under reflux in water bath. After addition of NaHCO₃ (until the evolution of CO₂ gas) the reaction mixture was filtrate. The filtrate was poured into water and extracted with petroleum ether. The combine organic layers were dried over Magnesium sulphate and the solvent was removed [8-9].

SPECTRAL ANALYSIS DATA

5- Amino Salicylic acid: ¹H-NMR: δ = 8.585 (Singlet 1H, -COOH), δ = 8.077 (1H, -OH phenolic) δ = 2.412 (Singlet, 2H, (NH₂ Primary amine), δ = 6.431 - 7.531 (Multiple 6H, Aromatic H). **IR Spectra:** (KBr) 3090 (N-H), 3160 (O-H), 2850 (C-H), 1650 (C=O), 1370-1600 (C=C & C-N). ¹³C-NMR: (Solvent CDCl₃) δ = 176.05 (-COOH), δ = 112.10 - 138.27 (Ar-C), and δ = 170.01 (-C=O).

(A) Methyl Derivatives: ¹H-NMR: δ = 10.226 (Singlet 1H, -OH), δ = 3.075 (Singlet, 2H, (-NH₂) Primary amine), δ = 6.831 - 7.279 (Multiple 6H, Ar-H), δ = 3.937 (Singlet 3H, -CH₃).

IR Spectra: (KBr) 3320 (N-H), 3200 (O-H), 2880-2920 (C-H), 1760 (C=O), 1240 (C-O), 1600, 1620, 1550 (C=C & C-N). ¹³C-NMR: δ = 59.60 (-CH₃), δ = 112.10 - 154.10 (Ar-C), and δ = 170.01 (-C=O).

(B) Ethyl Derivatives: ¹H-NMR: δ = 10.310 (Singlet 1H, -OH), δ = 3.457 (Singlet, 2H, (-NH₂) Primary amine), δ = 6.823 - 7.281 (Multiple 6H, Ar-H), δ = 4.424 (Quartet 2H, -CH₂) δ = 1.400 1.436 (Triplet 3H, CH₃). **IR Spectra:** (KBr) 3320 (N-H), 3210 (O-H), 2880 - 2920 (C-H), 1880 (C=O), 1280 (C-O), 1620, 1550 (C=C & C-N). ¹³C-NMR: δ = 14.21 (CH₃), δ = 61.31 (-CH₂), δ = 112.42 - 154.10 (Ar-C), δ = 170.01 (-C=O).

(C) N- Propyl Derivatives: ¹H-NMR: δ = 10.315 (Singlet 1H, -OH) δ = 3.484 (Singlet, 2H, (NH₂ Primary amine), δ = 6.820 - 7.279 (Multiple 6H Ar- H), δ = 4.277 - 4.309 (Triplet 2H, -CH₂), δ = 1.764 - 1.851 (Multiple 2H, -CH₂) and δ = 1.024 - 1.061 (Triplet 3H, -CH₃). **IR Spectra:** (KBr) 3360 (N-H), 3230 (O-H), 2890 - 2930 (C-H), 1870 (C=O), 1295 (C-O), 1640, 1690 (C=C & C-N) and 725 - 720 (C-C-C). ¹³C-NMR: δ = 10.49 (CH₃), δ = 21.97 (-CH₂), δ = 66.81 (-CH₂) δ = 112.45 - 154.87 (Ar-C), δ = 170.06 (-C=O).

(D) Iso Propyl Derivatives: ¹H-NMR: δ = 10.387 (Singlet 1H, -OH) δ = 3.484 (Singlet, 2H, (-NH₂)), δ = 6.790 - 7.279 (Multiple 6H, Aromatic H), δ = 5.205 - 5.298 (Multiple 1H, >CH) and δ = 1.359 - 1.375

(Doublet 6H, -CH₃). **IR Spectra:** (KBr)3360 (N-H),3230 (O-H), 2900-2940 (C-H stretch Ar -C),1340-1360 (C-H),1880(C=O),1300 (C-O),1640-1690(C=C & C-N),1280 - 1700 (C-C_v) and 1360-

1340(C-H_b). **¹³C-NMR:** δ = 21.84 (-CH₃), δ= 69.05 (-CH) δ =112.75 – 154.81 (Ar-C), δ = 169.56 (-C=O).

Table -1
Analytical Data and Some Physical Parameters of Compounds

Molecular formula of Derivatives	Name of Derivatives	Colour	Molecular weight	Yield %	Analysis (found/%calculated)			M.P.
					%C	%H	%N	
C ₇ H ₇ NO ₃	5-amino salicylic acid	Light pink	153	--	---	--	--	151
C ₈ H ₉ NO ₃	Methyl Derivative	Reddish brown	167	77.70	56.64 57.48	5.34 5.38	7.90 8.38	96
C ₉ H ₁₁ NO ₃	Ethyl Derivative	Reddish brown	181	72.37	58.45 59.66	6.02 6.07	7.49 7.73	114
C ₁₀ H ₁₃ NO ₃	N-propyl Derivative	Reddish black	195	68.75	60.50 61.53	6.17 6.72	6.98 7.17	---
C ₁₀ H ₁₃ NO ₃	Iso-propyl Derivative	Reddish black	195	62.55	60.50 61.53	6.17 6.72	6.98 7.17	---

ANTIMICROBIAL ACTIVITY

The newly Synthesized compounds A, B, C and D were screened for their antimicrobial activity by Agar diffusion method [10]. All the synthesized new titled derivatives were evaluated for the antimicrobial activity by E.coli, S. aureus, B. subtilis and S.typhi by measuring the zone of inhibition in mm. The activities were performed at a concentrate of 50µg / ml. Streptomycin sulphate (20µg/ml) was used as a standard drug for antimicrobial activity respectively. Alcohol was used as solvent control for antimicrobial activities. For antimicrobial activity both the derivatives (C & D) show antibacterial activity and show maximum inhibitory activity against E.coli. Result of sensitivity against S.aureus of derivative - D is

maximum while derivative - A show poor activity. S. typhi was highly sensitive to derivative -D while derivative - A showed very poor inhibition against it. B. subtilis was found good sensitive to derivative - D. This organism was not influenced by derivative -B. Derivative – A showed very poor inhibition of bacteria except for B. subtilis. Derivative – B showed inhibition of both positive and negative gram bacteria but could not inhibit sporulation bacteria. The assay of bacterial sensitivity was conducted under standard conditions of antibacterial assay technique (Methods in microbiology, A/P, 1978). The results were averaged from the duplicate plates of the concerned set of experiment.

Table – 2
Antimicrobial activity data of derivatives.

Derivatives	Diameter of zone of inhibition in mm.			
	E. coli	S. aureus	B. subtilis	S. Typhi
Methyl derivative	11	11	15	12
Ethyl derivative	18	15	----	17
N-propyl derivative	19	20	18	17
Iso-propyl derivative	20	22	16	18
Streptomycin sulphate (std. drug)	11	11	11	11

II. RESULT AND DISCUSSION:

A newly series of 5 - amino salicylic acid derivatives were reported in this paper. The target

derivatives compounds were synthesized by reflux method, The structure of the newly synthesized derivatives compounds have been elucidated on the

basis of Elemental, $^1\text{H-NMR}$, $^{13}\text{C-NMR}$, IR-Spectra and antimicrobial activities.

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