

# A Review on Advance Nano Sponge as Emerging Drug Delivery System

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Submitted: 15-10-2022

Accepted: 31-10-2022

## ABSTRACT

In recent years, drug targeting to the site of impact is a big problem, to overcome this problem, several new targeted drug delivery methods have been invented. Nano sponges are one such invention, these sponges are smaller in size than viruses and contain voids filled with various drugs. This sponge can enter the systemic circulation and travel all over the body until it interacts with a specific target site and binds to the surface and begins to release the drug in a controlled manner. An important feature of these sponges is their ability to dissolve in water and exert an effect on drugs with low solubility. This review focuses on preparation methods and applications of nanosponges in the field of drug delivery. Efficient, targeted drug delivery systems have been a long-standing dream, but have largely been frustrated by the complex chemistry involved in developing new systems. The invention of nano sponges has become an important step towards solving these problems.

**Keywords** nano-sponge, controlled release, poorly soluble drug, scaffold.

## I. INTRODUCTION

In the recent year's identification of novel chemical entities as well as its application to the detection and treatment of a wide range of diseases, nanotechnology has recently attracted a lot of researcher's attention. There has been significant advancement in the field of nanomedicine, including the development of novel drug delivery systems like nano-emulsion, nanoparticles, nanosponges, and nanosuspension. Nano-sponges, one such novel drug delivery system, offers a multiple benefit over competing approaches, including improved drug bioavailability. As a by-product known as nanomedicine, it has also had a favourable impact on the healthcare industry. As a result of advances in nanotechnology, specialised drug delivery systems have been created to achieve specific goals. Nano-sponges are similar in size to viruses. Due to the small size it quickly penetrates the skin. It's a brand-new kind of material with nanometer-wide cavities comprised of incredibly tiny particles. Drugs categorised under BCS class II

and IV that are poorly water soluble could potentially be placed into nano-sponges along with lipid-soluble drug molecules, improving their solubility<sup>[1,2]</sup>. One of the benefits of nano-sponge is that it is a three-dimensional scaffold (backbone) or network of polyester that can naturally decay. These polyesters are mixed with a cross-linker in a solution to create nano-sponges. Biodegradable polyester is frequently utilised in this condition, as it breaks down gradually in the body. They will distribute through a matrix of additives, lubricants, diluents, and anti-caking agents when taken orally, where they will be more effectively utilised to tablet and capsule formulation<sup>[3]</sup>.

## Difference between Nanoparticles and Nanosponge

The thin line separating nanoparticles from nanosponges is the difference in size and porosity. While nanosponges' overall size can range from micrometre to micrometre and is normally less than 5 m, nanoparticles have holes that are nanometer in size. Nanoporous nanoparticles/microparticles have been described as nanosponges on a number of occasions. The hydrophobic and hydrophilic groups that are present provide nanosponges a structure with a variety of domains. [5]

## Advantages of Nanosponges

- Improve solubility of lipophilic drugs in water.
- Protection of molecules and development of drug delivery systems for different routes of administration.
- They mix with water and serve as fluid carriers.
- To mask unpleasant tastes.
- Chemical linkers allow NS to bind specifically to the target site.
- NS engineering originates from the availability of relatively simple chemistry of polyesters and linker peptides.

## Disadvantages of nanosponges

- Nanosponges have the ability to encapsulate small molecules that are not suitable for larger molecules.

### Special Properties of Nanosponges

- Nanosponges have a range of sizes (1  $\mu\text{m}$  or less) with tunable polarization of the voids. Nanosponges of specific size and tunable polarity can be synthesized by varying the ratio of coupling agent to polymer.<sup>[4]</sup>
- Depending on the process conditions, they can be either para-crystalline or crystalline. The crystal structure of nanoparticles plays a very important role in their complexation with drugs. The drug loading capacity of nanosponges depends on the degree of crystallization. Para-crystalline nanosponges have shown different drug loading capabilities.<sup>[5]</sup>
- They are non-toxic, porous particles that are insoluble in most organic solvents and stable at high temperatures up to 300°C.<sup>[6]</sup>
- Nanosponges because the formulations are stable in the pH range from 1 to 11 and temperatures up to 130 °C.<sup>[7]</sup>
- They form clear, impure suspensions in water and can be recovered by simple thermal desorption.

### Chemicals used for the synthesis of nanosponges Polymers

Hyper cross-linked Polystyrenes, Cyclodextrins and its derivatives like Methyl  $\beta$ -Cyclodextrin, Alkylloxycarbonyl Cyclodextrins, 2-Hydroxy Propyl  $\beta$ - Cyclodextrins and Copolymers like Poly(valerolactoneallylvalerolactone) & Poly(valerolactoneallylvalerolactoneoxepanedione) and Ethyl Cellulose & PVA

### Cross linkers

Diphenyl Carbonate, Diarylcarbonates, Diisocyanates, Pyromellitic anhydride, Carbonyldiimidazoles) Epichloridrine, Glutaraldehyde, Carboxylic acid dianhydrides, 2,2bis(acrylamido) Acetic acid

### Factors affecting the formulation of NanoSponge Polymer Type

The formation as well as the performance of nanosponge depends on the selection of the appropriate polymer. The cavity or pore size of the nanosponge must be able to accommodate drug molecules of the appropriate size.<sup>[13]</sup>

### Drugs

- Molecular weight must be between 100 and 400 Daltons.

- The structure of the drug molecule should not contain more than five condensate rings.
- Water solubility should be less than 10 mg/ml.
- The melting point must be below 250°C.<sup>[15]</sup>

### Temperature

Temperature variations can affect drug complexation. Increasing temperature reduces the stability of NanoSponge complex which is clearly possible because NanoSpongedrug interaction force, Vander Waals force and hydrophobic force can be reduced with increasing temperature.<sup>[14]</sup>

### METHOD OF PREPARATION

#### Nanosponges made from hyper cross-linked $\beta$ -cyclodextrins

super-crosslinked

Nanosponge is made from a material that generates non-porous molecules as a carrier called cyclodextrin for drug delivery. These cyclodextrins are a super-crosslinking agent that forms many networks in the nano network, or can even be spherical with many protein channels, networks, pores, etc. These crosslinking agents stabilize sponges with specific surface charged densities, porosity, and pore sizes of based on the molecules they contain. Crosslinking agent help keep Nanosponges at different acidic and neutral pH values.<sup>[8]</sup>

#### Solvent emulsification method

Main polymers used in this method are ethyl-cellulose and polyvinyl alcohol in variable proportions. The dispersed phase was formed by adding ethyl cellulose and the available drug was dissolved in 20 ml of dichloromethane. The continuous phase dropwise addition was prepared by dissolving polyvinyl alcohol in 150 ml of distilled water. The mixture was then stirred at 1000 rpm for about 2 h. The obtained Nano sponges will be collected, filtered and dried in an oven for about 1 day and then stored in a desiccator.<sup>[9]</sup>

#### Solvent method

The polymers used above can be used with suitable polar aprotic solvents such as dimethylformamide, dimethyl sulfoxide and mixed in proportion. Then, to this mixture, available crosslinkers are added in a ratio of 4:16. The temperature was maintained from 10 °C for the reaction of the polymers for 2 days. Most carbonyl crosslinkers (dimethyl carbonate and carbonyl diimidazole) are used. When the reaction was complete, the product was kept cooled to room temperature, then the mixture was added with distilled water for recovery and filtration in an air

furnace and purification was performed using a Soxhlet apparatus with the addition of water. add ethanol for additional extraction. Again, opt for vacuum drying and mechanical beating to obtain a uniform white powder.<sup>[10]</sup>

#### Ultrasonic-assisted synthesis

In this process, nanosponges can be obtained using polymers with carbonyl cross-links under solvent-free conditions and stored for sonication. These Nano-developed sponges will have a uniform spherical size. Mix polymer and crosslinking agent in sufficient quantity and collect in bottle. The flask was filled with water and heated to 90°C for sonication. The mixture was kept for 5h for continuous negatives. The mixture was then cooled, and the product was washed with distilled water and purified using a Soxhlet extractor using ethanol. The final product obtained was dried at 25°C and the white powder was recovered and stored away from moisture.<sup>[11]</sup>

#### NanoSpongedrugloading

Nanosponges formulated for drug delivery must first be pretreated to achieve an average particle size less than 500nm. The Nanosponges are then suspended in water for a period of time and treated with ultrasonic waves to avoid the formation of agglomerates. The product suspension obtained was centrifuged to obtain the colloid. The obtained supernatant was separated and the sample was dried by lyophilization.<sup>[12]</sup>

Alternatively, aqueous suspensions of Nanosponges are prepared and dispersed with continuous stirring for a specific period of time. NanoSpongesolid crystals are obtained by evaporating the solvent or by lyophilization. The crystalline structure of the nanoporous foam plays a very important role in the drug complexation process. The drug loading capacity in crystalline nanosponges is higher than in crystalline forms. In nanosponges containing porous or crystalline structure, drug loading occurs as mechanical mixing rather than formation of an inclusion complex.

### APPLICATIONS OF NANOSPONGES

#### Nanosponges as sustained release system

Acyclovir is one of the widely used antiviral agents for the treatment of herpes simplex virus infections. Its absorption in GIT is slow and incomplete and very variable. The in vitro release profile of acyclovir from different Nanosponges indicates sustained release. The percentage of acyclovir release from Nano carb sponge and Nanosponge after 3 hours of use is about 22% and 70%. The drug was not absorbed on the surface of the NanoSponges as no initial explosive effect was observed.<sup>[16]</sup>

#### Nanosponges in Solubility Enhancement

Itraconazole is a BCS drug with limited solubility and low bioavailability. Therefore, the application of nanosponges

improved the drug solubility by more than 27-fold. Solubility was found to be in excess of 55-fold, when copolyvidonum was added as a carrier component. Either by masking the hydrophobic groups of itraconazole, increasing drug wetting, or reducing drug crystallinity, nano-slits improved drug solubility.<sup>[17]</sup>

#### Nanosponges in solubility enhancement

Itraconazole is a BCS class II drug which has a dissolution rate limited poor bioavailability. Thus the application of nanosponges improved the solubility of the drug more than 27-fold. The solubility was found to be exceeded to 55fold, when copolyvidonum was added as a Supporting component. Either by masking the hydrophobic groups of itraconazole, by increasing the wetting of the drug or by decreasing the crystallinity of the drug nanosponges improve the solubility of the drug.<sup>[17]</sup>

#### NanoSponges in Drug Delivery

Nanosponges can be formulated in different dosage forms such as topical, injection, aerosol, tablets and capsules. Telmisartan (TEL) is a drug with limited bioavailability. TEL was incorporated into the nanosponge formulation. The saturation solubility and in vitro solubility of the TEL- $\beta$ -CD complex were compared with that of the conventional TEL and TEL nanosponge complexes. The highest in vitro drug solubility and release were observed in inclusion complexes prepared from nanosponge and NaHCO<sub>3</sub>. Paclitaxel is a poorly water-soluble antineoplastic drug.  $\beta$ -CD-based nanosponges are an alternative to the classical cremophore formulation because cremophore reduces the issue penetration of paclitaxel. The biological effects of paclitaxel in vitro are significantly enhanced by the nanosponge formulation. Econazole nitrate is an antifungal agent used for skin infections and fungal skin diseases. Absorption is negligible when econazole is applied to the skin. Therefore, econazole nitrate nanofoams were generated by solvent diffusion and loaded as hydrogels.<sup>[18]</sup>

#### Nanosponges in Enzyme Immobilization

Nanosponges have been widely used for enzyme stabilization. CD-NS exhibits a much higher integration constant than CD and is suitable for supporting enzyme immobilization. They help maintain the catalytic capacity and stability of immobilized enzymes. Enzyme immobilization is important for enzyme recycling and facilitates separation and recovery of the formed products as well as increases the thermal stability and biocatalyst activity. Boscolo et al. also investigated the high catalytic efficiency of some Pseudomonas fluorescens lipases adsorbed on cyclodextrin-

based nanosponge. Lipases are widely used to catalyze triacylglycerol hydrolysis and transesterification reactions, with many industrial applications.<sup>[19]</sup>

#### **Nanospheres for protein delivery**

A major obstacle in protein formulation development is maintaining the original protein structure both during formulation and during long-term storage. Swaminathan et al. investigated novel expandable polyananospheres based on cyclodextrin. Through water absorption studies, they observed very good swelling for 72 hours. Bovine serum albumin was used as the template protein and incorporated into the prepared nanosponge. Improved swelling properties as well as increased protein stability were observed. At physiological pH, the lactone ring opens and develops an inactive carboxylate form. The fusion of camptothecin in the nanospheres results in a sustained release profile in the active form, which prevents hydrolysis of the lactone form and leads to improved stability.<sup>[20]</sup>

#### **Cancer Therapies<sup>[21]</sup>:**

Nanospheres can be used as an anti-tumor drug delivery device for cancer drugs. The method was three to five times more effective than direct injection of the drug in reducing tumor growth, the researchers said. The small nanospheres are encapsulated with a dose of the drug and display a binding peptide that binds to cell surface targets by radiation to the tumor. They stick to the surface when the sponge comes into contact with tumor cells and are stimulated to expel their charge. The benefit of targeted drug delivery allows for more accurate diagnosis at the same dose and fewer health problems. Studies have been conducted as a sponge filler in animals with paclitaxel.

#### **Antiviral application<sup>[22]</sup>:**

In all cranial, nasal, pulmonary therapeutics, nanospheres may be beneficial. To capture RTI-infecting viruses, including respiratory membrane cell viruses, influenza viruses, and hantaviruses, nanocarriers can selectively deliver antiviral drugs or small interfering RNAs (siRNA) to the nasal and lung epithelium. They can also be used for HIV, HBV, and HSV. Zidovudine, saquinavir, interferon- $\alpha$ , acyclovir (Eudragit-based) are drugs commonly used only as a nano-delivery system.

#### **Gas Encapsulation<sup>[22]</sup>:**

Cyclodextrin-dependent Nanosphere is used to form complexes consisting of 3 separate gases, namely 1-methylcyclopropane, oxygen and carbon dioxide. For some biomedical applications, complexation of oxygen or carbon dioxide can be useful. In particular, oxygen-

filled nanosphere can deliver oxygen to hypoxic tissues in various diseases. Due to its high nature, extremely porous; As an efficient gas carrier, Nanosphere has also been explored. The composition of nanosphere shows the ability to monitor the accumulation and release of oxygen. They could be a valuable tool to deliver some important gases in the future.

#### **Other uses of nanospheres**

Cyclodextrin-based nanospheres can strongly bind organic molecules and extract them even at the low water levels. This same concept could be beneficial when selectively combining a polymer and a crosslinker to remove bitter components from grape juice. Three-dimensional nanospheres for proteomics applications can play an important role in peptide fractionation. For gases such as oxygen and carbon dioxide, nanospheres can be used as a transport medium.

#### **Oral distribution**

This complex can be dispersed in a matrix of diluents, excipients, lubricants and anti-caking agents for oral administration in capsule or tablet form. Acetylsalicylic acid, a nonsteroidal anti-inflammatory drug (NSAID) classified as a Class IIIB CSDrug, is used to form nanospheres for use in an oral drug delivery system.<sup>[23]</sup>

#### **Topical use**

The components of Nanosphere can be applied topically as a gel or cream. It is believed that the nanoparticles are loaded with resveratrol.<sup>[24]</sup>

#### **Characterization of the nanoparticles**

Methods for the characterization of complexing drugs/nanospheres are listed below: Solubility studies Inclusion complexes are a technique for determining drug solubility and bioavailability. This technique is the most widely discussed technique for the analysis of nanosphere complexes. The degree of completion can be known by plotting the solubility of the phases. Solubility studies are performed to access the pH of the drug, outline solubility, and to evaluate the factors affecting drug solubility.<sup>[16]</sup>

#### **X-ray diffraction studies**

For the solid state, X-ray powder diffraction measurement can be used to determine the complexity of impurities. When the drug molecule is liquid and the liquid itself has a zero-diffraction pattern, the diffraction pattern of a newly formed substance will be markedly different from the diffraction pattern of an incomplete nanosphere. This difference in the diffraction pattern indicates the formation of a



complex. When the drug compound is a solid, it is necessary to compare the diffraction pattern of the complex and that of the mechanical mixture of drug molecules and polymers. The diffraction pattern of a physical mixture is usually the sum of the forms of each component, while the diffraction patterns of a complex are obviously different for each component and result in a new solid phase with different diffraction patterns. Together, the diffraction peaks for mixtures of compounds are useful in determining chemical decomposition and complex formation. Complex formation of the drug with nanosponge changes the diffraction pattern and also changes the crystalline nature of the drug. The complex formation leads to the emphasis of existing peaks and the displacement of certain peaks.

### Microscopic studies

Microscopic studies of drug/nanofoam can be performed using scanning electron microscopy and transmission electron microscopy. The inclusion complex formation represented by the difference between the crystallization and product states was observed under the electron microscope.

### Thermodynamic method

If the changes occur in the drug molecule or the particles change earlier than the thermal decomposition of the nanofoam, it can be determined by thermochemical method. The drug particle changes can be melting, evaporation, oxidation and decomposition and macromolecular changes. Changes in the drug molecules suggest the formation of a good complex.

### Particle size and dispersion Polydispersity

Particle size was determined by dynamic light scattering using 90Plus particle size determination software. Dynamic light scattering (DLS) is defined as a technique used to know the size distribution profiles of nanoparticles. Finally, the final particle diameter and dispersion index (PDI) can be found.

### Thin layer chromatography (TLC)

TLC can be defined as a technique that can be used to separate non-volatile or volatile mixtures. In this technique, if the R<sub>f</sub> value of a particular drug molecule is within an acceptable range, it is useful in recognizing the formation of a complex between the drug and the nanosponges.

### Infrared Spectroscopy

Interactions between nanosponges and solid-state drugs can be determined by infrared spectroscopy. The nanosponge bands may change slightly upon complex formation. Few guest molecules are bound in the complex with a fraction less than 25%, so the drug spectrum

can be easily masked by the nanosponge spectrum. This technique is not suitable for determining inclusion complexes compared to other methods<sup>[17]</sup>.

### Charging efficiency

The loading efficiency of nanosponge can be determined by estimating the drug loaded into the nanosponge using UV spectrophotometer and high-performance liquid chromatography for nanoparticles. The charging efficiency of nanosponges can be calculated using the following formula.

$$LE = \frac{\text{Actual drug content in nanosponges}}{\text{Theoretical drug content}} \times 100$$

## II. CONCLUSION

Nanosponges can deliver both hydrophilic and hydrophobic drugs, and they are drug delivery systems that enable controlled and predictable drug release to a target site, thereby improving efficiency and bioavailability. They can be formulated in many different formulations such as topical, oral and systemic.

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